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- Sulfur
dibromide has a
dot structure
that begins with
the S atom in the
center On the
left and right
sides are a
singly bonded
Br atom On the
unbonded sides of
each atom is a ?
single pair of
dots' 'SHORT

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- Iodine is a
diatomic molecule
so its molecules
are paired as I_2
Iodine has 7
electrons in its
outer shell so 1
electron is

shared by each
atom' 'Worksheet

Lewis Dot

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- For each of the
following draw
the Lewis Dot
Structure give
the electron
arrangement E A
and the molecular
geometry M

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- Heating and
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Notice that in
general the

temperature goes

up the longer the heating continues'

'Worksheet Lewis Dot Arkansas State University

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- For each of the following draw the Lewis Dot Structure give the electron arrangement E A and the molecular geometry M

G' ***'Heating and Cooling Curves The Basics AP Chemistry***

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- *Heating and Cooling Curves of Substances Click here for Calculations and Heating Curves Notice that in general the temperature goes up the longer the heating continues'*

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- *Iodine is a diatomic molecule so it s molecules are paired as I₂ Iodine has 7 electrons in it s outer shell so 1 electron is shared by each*

atom'

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