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chemfiesta com](http://www.chemfiesta.com) Solutions for the**

Balancing Equations Practice

Worksheet 1 $2 \text{NaNO}_3 \rightarrow 2 \text{Na}_2\text{O} + 2 \text{N}_2 + 3 \text{O}_2$ $\text{PbO} + \text{Pb} + \text{NO}_3$

$2 \text{Na}_2\text{O} + 2 \text{H}_2\text{O} \rightarrow 2 \text{NaOH} + \text{H}_2$ $6 \text{AgI} + \text{Fe}_2(\text{SO}_4)_3 \rightarrow 2 \text{FeI}_3 + 3 \text{Ag}_2\text{SO}_4$

$3 \text{Ag}_2\text{CO}_3 \rightarrow 3 \text{Ag}_2\text{O} + 3 \text{CO}_2$ $2 \text{C}_2\text{H}_4\text{O}_2 + 2 \text{O}_2 \rightarrow 2 \text{CO}_2 + 2 \text{H}_2\text{O}$

H₂O"Dealing with the chemical

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reactions The central problem of
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Balance the following chemical

equations"*Skills Worksheet Concept*

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law of conservation of mass states that

in ordinary chemical or physical

changes mass is neither created or'

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Balancing Chemical Equations

?Types of Reactions ?Predicting

Products ?Predicting States of

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Test Review List the 5 types of
chemical reactions and classify each
of the reactions in part 2 above
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like"Unit 7 Chemical Equations and Reactions

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of the most important topics in the
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Worksheet 1 2 NaNO_3 PbO Pb NO_3 2

Na_2O 2 6 AgI Fe_2 CO_3 3 2 FeI_3 3

Ag_2CO_3 3 $\text{C}_2\text{H}_4\text{O}_2$ 2 O_2 2 CO_2 2

H_2O '

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science world'**

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1 10**

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Chemical Equations? 1 1 10'**

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The law of conservation of mass
states that in ordinary chemical or
physical changes mass is neither
created or'**

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initial unbalanced Balancing
equations with reactants and
products"Chemical Equations**

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reacting and the substances that are
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oxygen O_2 to form dinitrogen**

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